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## Improvement of Eucken correction for high-temperature gases

**V.Istomin, E.Kustova, M.Mekhonoshina**

*Saint Petersburg State University*

- Simple engineering formulas are necessary for the calculation of the transport coefficients in high-temperature gases.
- Transport properties of electronically excited gases are actively studied during the last decade due to their importance in re-entry problems.
- For gases with electronic excitation simple relation for thermal conductivity coefficient hasn't been assessed up to now.

- 1913 – E.Eucken on the basement of mean free path concept proposed an approximate analytical expression for the thermal conductivity coefficient.
- 1957 – Using the Chapman-Enskog method Hirschfelder introduced a correction to refine the Eucken formula.
- 1990-2000 – Transport properties of electronically excited gases were investigated by Capitelli, Bruno, Kustova, Istomin and others.
- Applicability of Eucken formula to high temperature conditions, when electronic excitation is important, was not discussed in the literature except some papers.

- Using methods of the kinetic theory
  - to develop an accurate model for calculating the thermal conductivity coefficient of gases with excited rotational, vibrational and electronic degrees of freedom,
  - to modify Eucken formula for gases with electronic excitation.
- Using the accurate model to estimate the limits of applicability of simplified Eucken and Hirschfelder relations in the temperature range 200 – 25000 K.

If the molecules do not have the internal energy, the following approximate relation can be used:  $\lambda = \frac{5}{2}\eta c_v$ .

For particles with internal energy, on the basis of the mean free path consideration, Eucken suggests to separate  $\lambda$  into two parts:

$$\lambda = \lambda_{tr} + \lambda_{int} = (c_{v,tr} f_{tr} + c_{v,int} f_{int}) \eta,$$

factors:

$$f_{tr} = \frac{5}{2}, \quad f_{int} = 1.$$

Components of the specific heat due to the translational motion and internal energy of the molecules: factors:

$$c_{v,tr} = \frac{3k}{2m}, \quad c_{v,int} = c_v - c_{v,tr}.$$

Using the Chapman-Enskog theory the Eucken formula can be improved. Hirschfelder has shown that the thermal conductivity can be expressed as follows:

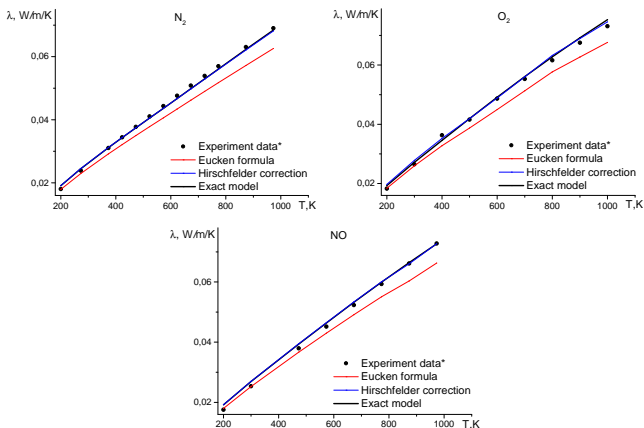
$$\lambda = \left( \frac{5}{2} c_{v,tr} + \frac{\rho D}{\eta} c_{v,int} \right) \eta.$$

The mean value  $f_{int} = \frac{\rho D}{\eta} = 1.328$  provides a good approximation for the Lennard-Jones model.

This expression is more accurate than the Eucken formula, but for some cases it gives inaccuracy exceeding the experimental uncertainty.

- We consider a non-equilibrium gas mixture flow under condition of fast relaxation of internal energy and slow chemical reactions (one-temperature approximation)
- Algorithm for calculating the transport coefficients is based on zero- and first-order approaches of the modified Chapman-Enskog method.
- Thermal conductivity coefficient expressed as  $\lambda' = \frac{k}{3} [\mathbf{A}, \mathbf{A}] = \lambda_{tr} + \lambda_{int}$  and describes in one-temperature approximation transfer of translational  $\lambda_{tr}$  and all  $\lambda_{int}$  internal energy types.
- In this work we consider gases O, N, O<sub>2</sub>, N<sub>2</sub>, NO
- We take into account 170 and 204 excited electronic states of the nitrogen and oxygen atoms
- 5, 7 and 4 electronic terms of the nitrogen, oxygen and nitric oxide molecules
- Temperature range 200 – 20000 K

Figure : Thermal conductivity coefficient as a function of T



\*Uribe F.J., Kestin J. and Mason E.A. Thermal conductivity of nine polyatomic gases at low density // J. Phys. Chem. Ref. Data, V. 19.

5. P. 1123-1136, 1990.



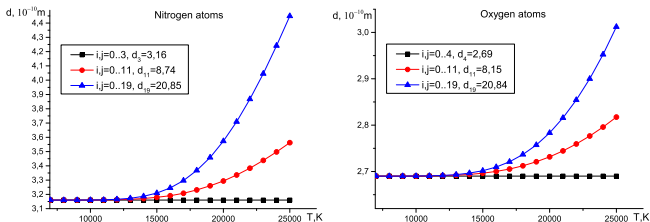
We introduce the effective diameter:

$$d_{eff}^2 = \frac{1}{Z_{int}^2} \sum_{nm} s_n s_m \exp\left(\frac{-\varepsilon_n - \varepsilon_m}{kT}\right) \left(\frac{d_n + d_m}{2}\right)^2,$$

and obtain a simplified averaging operator:

$$\langle F \rangle = \left(\frac{kT}{\pi m}\right)^{1/2} \pi d_{eff}^2 \int F(\gamma) \gamma^3 \exp(-\gamma^2) d\gamma,$$

Figure : Effective diameters as functions of T



The average value for the factor  $f_{tr}$  equal  $\frac{5}{2}$  in the whole temperature range.

Figure : Eucken factor  $f_{int}$  for air species as a function of T

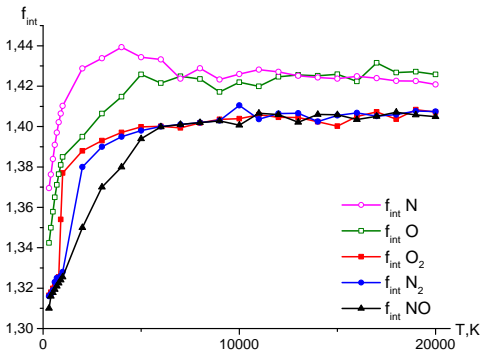
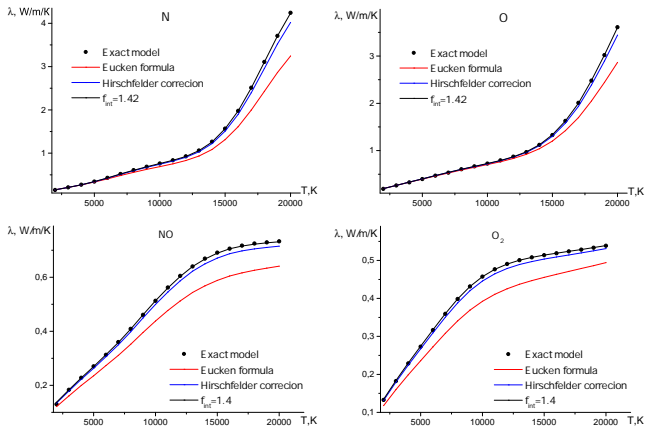


Figure : Thermal conductivity coefficients of air species as functions of T



**Table :** The values of the Eucken factor  $f_{int}$  which provide the best accuracy.

c	$T < 20000$ K		
<i>N</i>	1.42		
<i>O</i>	1.42		
c	$T < 1000$ K	$1000 < T < 5000$ K	$5000 < T < 20000$ K
<i>NO</i>	1.328	$-2 \cdot 10^{-9} T^2 + 3 \cdot 10^{-5} T + 1.2988$	1.40
<i>O<sub>2</sub></i>	1.328	$-3 \cdot 10^{-9} T^2 + 3 \cdot 10^{-5} T + 1.3436$	1.40
<i>N<sub>2</sub></i>	1.328	$-8 \cdot 10^{-9} T^2 + 6 \cdot 10^{-5} T + 1.2783$	1.40

Our model provides satisfactory accuracy for temperatures lower than 20000 K.

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Thank you for your attention!